

# Schistosomicidal and Antioxidant Flavonoids from *Astragalus englerianus*

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## Key words

- *Astragalus englerianus*
- Fabaceae
- flavonoids
- schistosomicidal
- antioxidant activity

## Abstract

*Astragalus englerianus* is a close relative of the traditional Chinese medicine plant Radix Astragali (Huang-qi) and is mainly distributed in Yunnan. It has been traditionally used as a substitute of “Huang-qi” for reducing fatigue and enhancing immunity by local folks. A phytochemical study of the methanol extract of the roots led to the isolation of three new flavonoids including one aurone (**1**) and two chalcones (**2** and **3**), as well as two known flavonoids (**4** and **5**). Their structures were elucidated based on the analyses of extensive spectroscopic data and comparison of their physicochemical properties. This is the first report on the occurrence of  $\beta$ -hydroxydihydro-

chalcone, 2',5'-dioxxygenchalcones, and 2',5'-dioxxygenaurone in the genus *Astragalus*. All the isolated compounds were tested *in vitro* for their schistosomicidal and antioxidant activities. Compounds **2** and **4** showed schistosomicidal activities with worm mortality rates of 100% within 12 h in a drug-containing (0.70 and 0.77 mM, respectively) RPMI 1640 medium. Compounds **1** and **2** exhibited antioxidant activities in 2,2-diphenyl-1-(2,4,6-trinitrophenyl)hydrazyl free radical scavenging assays, with IC<sub>50</sub> values of 35.9 ± 1.1 and 12.2 ± 1.1  $\mu$ M, respectively.

**Supporting information** available online at <http://www.thieme-connect.de/products>

## Introduction

As an important traditional Chinese medicine, Radix Astragali (Huang-qi) has been used for more than two thousand years to enhance and regulate immune function and to promote metabolism. It has been verified to possess antifatigue, anti-anoxia, anti-radiation, and hepatoprotective activities [1,2]. Official materials used as Radix Astragali come from two *Astragalus* plants, *Astragalus membranaceus* and *Astragalus membranaceus* var. *mongholicus*. However, neither of these grow in Yunnan [3]. *Astragalus* plants distributed in China include 278 species, two subspecies, 35 varieties, and two forms [4]. Among them, at least 51 *Astragalus* species can be found in Yunnan, and most of them are used as substitutes for “Huang-qi” by local folks and display similar medicinal effects to Radix Astragali [1]. Although *Astragalus* resources in Yunnan are exceptionally rich, most *Astragalus* plants have not been systematically studied yet. Unfortunately, due to environmental changes and human influences, some *Astragalus* plants in Yunnan became

endangered during the past several decades. It is, therefore, necessary to perform urgent studies on *Astragalus* plants for both scientific research and plant preservation purposes. In this study, one Yunnan native and commonly used *Astragalus* plant, *Astragalus englerianus* Ulbr. (Fabaceae or Leguminosae), was selected for chemical and bioactive studies. This paper is the first report of natural products from *A. englerianus*.

## Results and Discussion

Three new flavonoids, (*Z*)-2',5'-dihydroxy-6-methoxyaurone (**1**), 2,2',5'-trihydroxy-4-methoxychalcone (**2**), and ( $\beta$ R)-2,2',5', $\beta$ -tetrahydroxy-4-methoxydihydrochalcone (**3**), together with two known flavonoids, (3R)-sativan (**4**) [5,6] and (6aR,11aR)-maackiain (**5**) [7], were isolated and identified from the methanol extract of the root (● Fig. 1).

Compound **1** was assigned the molecular formula C<sub>16</sub>H<sub>12</sub>O<sub>5</sub> (11 unsaturation degrees) from its HR-EI-MS. The IR spectrum showed absorption

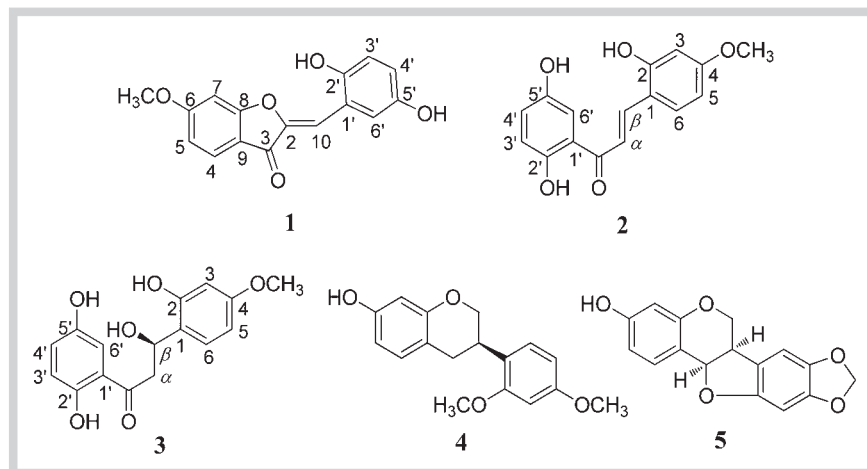
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**Fig. 1** Structures of compounds 1–5.

**Table 1** NMR spectroscopic data of compounds 1–3.

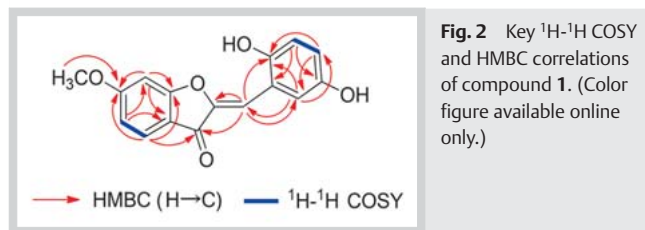
Position	1 <sup>a</sup>		2 <sup>b</sup>		3 <sup>c</sup>	
	$\delta_{\text{H}}$ (multi, <i>J</i> in Hz)	$\delta_{\text{C}}$	$\delta_{\text{H}}$ (multi, <i>J</i> in Hz)	$\delta_{\text{C}}$	$\delta_{\text{H}}$ (multi, <i>J</i> in Hz)	$\delta_{\text{C}}$
1				116.1		119.6
2		148.4		160.0		156.6
3		183.1	6.57 (d, 2.3)	102.3	6.42 (d, 2.5)	102.2
4	7.76 (d, 8.3)	126.1		164.4		162.2
5	6.75 (brd, 8.3)	113.1	6.53 (dd, 8.6, 2.3)	107.7	6.49 (dd, 8.56, 2.5)	106.0
6		167.9	7.75 (d, 8.6)	132.1	7.40 (d, 8.56)	128.8
7 (-CO-)	6.53 (brs)	97.3		194.8		195.5
8 ( $\alpha$ )		168.9	7.84 (d, 15.4)	118.3	3.02 (dd, 17.0, 13.5) 2.79 (dd, 17.0, 2.7)	44.5
9 ( $\beta$ )		115.9	8.26 (d, 15.4)	142.2	5.65 (dd, 13.5, 2.7)	76.3
10	8.13 (s)	107.7				
1'		121.6		121.0		122.2
2'		152.4		157.8		157.5
3'	7.20 (overlap)	117.9	6.86 (d, 8.9)	119.5	6.94 (d, 8.9)	120.2
4'	7.24 (overlap)	120.8	7.13 (dd, 8.9, 2.9)	125.4	7.05 (dd, 8.9, 3.1)	125.9
5'		152.8		150.1		152.9
6'	8.42 (brs)	118.4	7.56 (d, 2.9)	115.4	7.23 (d, 3.1)	111.4
-OCH <sub>3</sub>	3.75 (s)	56.7	3.80 (s)	55.8	3.77 (s)	55.7
2'-OH			12.64 (s)			
-OH			9.48 (brs)			

<sup>a</sup> Recorded at 400 MHz for <sup>1</sup>H NMR and 100 MHz for <sup>13</sup>C NMR in pyridine-*d*<sub>5</sub>; <sup>b</sup> Recorded at 400 MHz for <sup>1</sup>H NMR and 100 MHz for <sup>13</sup>C NMR in acetone-*d*<sub>6</sub>; <sup>c</sup> Recorded at 400 MHz for <sup>1</sup>H NMR and 100 MHz for <sup>13</sup>C NMR in methanol-*d*<sub>4</sub>

bands for hydroxyl (3405 cm<sup>-1</sup>), conjugated carbonyl (1666 cm<sup>-1</sup>), conjugated double bond (3025, 1629 and 910 cm<sup>-1</sup>), and aromatic ring (3074, 1590, and 1501 cm<sup>-1</sup>) functional groups. The <sup>1</sup>H NMR spectrum of **1** showed an ABX coupling system with protons at  $\delta_{\text{H}}$  7.76 (1H, d, *J* = 8.3 Hz, H-4), 6.75 (1H, brd, *J* = 8.3 Hz, H-5), and 6.53 (1H, brs, H-7), and one unknown coupling system with signals at  $\delta_{\text{H}}$  8.42 (1H, brs, H-6'), 7.24 (1H, overlap, H-4'), and 7.20 (1H, overlap, H-3'). The <sup>1</sup>H NMR spectrum of **1** also exhibited one tri-substituted ethylene structural unit at  $\delta_{\text{H}}$  8.13 (1H, s, H-10) and one methoxyl at  $\delta_{\text{H}}$  3.75 (3H, s). Analysis of the <sup>13</sup>C NMR spectrum with the aid of the DEPT spectra revealed the existence of 16 carbons (Table 1). The spectroscopic data mentioned above indicated that compound **1** should be an aurone. Comparison of the spectroscopic data with those of aurones previously reported [8–10] suggested that compound **1** was an aurone with three oxygen-containing substituents.

The structure of **1** was determined by 2D-NMR analyses. The correlations from the <sup>1</sup>H-<sup>1</sup>H COSY spectrum suggested two segments (Fig. 2, bold). The HMBC correlations of methoxyl ( $\delta_{\text{H}}$  3.75, s) to C-6 ( $\delta_{\text{C}}$  167.9), of H-4 ( $\delta_{\text{H}}$  7.76, d, *J* = 8.3 Hz) to C-3 ( $\delta_{\text{C}}$  183.1), and of H-10 ( $\delta_{\text{H}}$  8.13, s) to C-3 and C-6' ( $\delta_{\text{C}}$  118.4), as well as the other correlations shown in Fig. 2, confirmed that compound **1** was an aurone with three oxygen-bearing substituents at C-6, 2' and 5'. Thus, compound **1** was identified as (*Z*)-2',5'-dihydroxy-6-methoxyaurone.

Compound **2** possessed the molecular formula C<sub>16</sub>H<sub>14</sub>O<sub>5</sub> (10 double bond equivalents) according to HR-El-MS. The IR spectrum showed absorption bands for hydroxyl (3395 and 3387 cm<sup>-1</sup>), conjugated carbonyl (1639 cm<sup>-1</sup>), conjugated *trans*-double bond (3020, 1611 and 987 cm<sup>-1</sup>), and aromatic ring (3059, 1588 and 1511 cm<sup>-1</sup>) functional groups. The <sup>1</sup>H NMR spectrum of **2** showed the signals of one methoxyl, two hydroxyls, two ABX coupling system hydrogens, and one double bond (Table 1), which con-

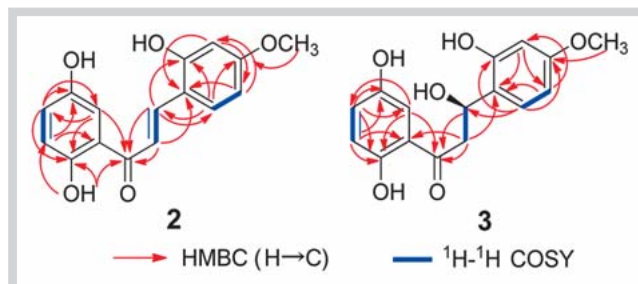


**Fig. 2** Key  $^1\text{H}$ - $^1\text{H}$  COSY and HMBC correlations of compound **1**. (Color figure available online only.)

firming the existence of a conjugated *trans*-double bond in **2**.  $^{13}\text{C}$  NMR and DEPT spectra displayed the existence of 16 carbons (Table 1) including one methoxyl, one carbonyl, and 14 olefinic carbons (including eight  $\text{sp}^2$  methines, and six  $\text{sp}^2$  quaternary carbons). After comparison of the spectroscopic data with those of known chalcones [8, 11], compound **2** was considered as a typical chalcone with four oxygen-containing substituents.

The correlations observed in the  $^1\text{H}$ - $^1\text{H}$  COSY spectrum suggested three fragments, H-5/H-6, H-3'/H-4', and H- $\alpha$ /H- $\beta$  (Fig. 3), which showed **2** having two ABX aromatic coupling systems. The  $^1\text{H}$  NMR spectrum of **2** showed a hydrogen-bonded hydroxyl proton at  $\delta_{\text{H}}$  12.64 (1H, s) that can be assigned to the 2'-OH group, which was further confirmed on the basis of the correlations between 2'-OH and C-2' ( $\delta_{\text{C}}$  157.8), C-3' ( $\delta_{\text{C}}$  119.5), and carbonyl ( $\delta_{\text{C}}$  194.8) in the HMBC spectrum. A correlation of methoxyl ( $\delta_{\text{H}}$  3.80, s) to C-4 ( $\delta_{\text{C}}$  164.4) in the HMBC spectrum indicated that the methoxyl was at the position of C-4. Correlations of H-6' ( $\delta_{\text{H}}$  7.56, d,  $J = 2.9$  Hz) to C-2' ( $\delta_{\text{C}}$  157.8), C-4' ( $\delta_{\text{C}}$  125.4), and carbonyl ( $\delta_{\text{C}}$  194.8), of H- $\alpha$  ( $\delta_{\text{H}}$  7.84, d,  $J = 15.4$  Hz) to C-1 ( $\delta_{\text{C}}$  116.1) and carbonyl ( $\delta_{\text{C}}$  194.8), and of H- $\beta$  ( $\delta_{\text{H}}$  8.26, d,  $J = 15.4$  Hz) to C-2 ( $\delta_{\text{C}}$  160.0), C-6 ( $\delta_{\text{C}}$  132.1), and carbonyl ( $\delta_{\text{C}}$  194.8) in the HMBC spectrum, as well as the other correlations depicted in Fig. 3, confirmed the chalcone skeleton and suggested that the other two hydroxyl groups were at C-2 and C-5', respectively. Thus, compound **2** was proposed as 2,2',5'-trihydroxy-4-methoxychalcone. Compound **3** had the molecular formula  $\text{C}_{16}\text{H}_{16}\text{O}_6$  (nine unsaturation degrees) on the basis of its HR-EI-MS. The IR spectrum showed absorption bands for hydroxyl (3405 and 3283  $\text{cm}^{-1}$ ), conjugated carbonyl (1671  $\text{cm}^{-1}$ ), and aromatic ring (3025, 1611, and 1513  $\text{cm}^{-1}$ ) functional groups.  $^1\text{H}$  NMR and  $^{13}\text{C}$  NMR spectral data of **3** (Table 1) were similar to those of **2** except for the disappearance of a disubstituted ethylene structural unit and the observation of an oxygen-bearing methine carbon at  $\delta_{\text{C}}$  76.3 and a methylene at  $\delta_{\text{C}}$  44.5. Further analyses of 2D NMR spectroscopic data ( $^1\text{H}$ - $^1\text{H}$  COSY and HMBC, Fig. 3), and the specific rotation data  $[\alpha]_{\text{D}}^{25} -18.82$  (c 0.13, MeOH) of compound **3** and comparison with those of elatadihydrochalcone [12] allowed the spatial structure to be established. Therefore, compound **3** was deduced as (*BR*)-2,2',5', $\beta$ -tetrahydroxy-4-methoxydihydrochalcone. The two known compounds were identified as (*3R*)-sativan (**4**) [5, 6] and (*6aR*,*11aR*)-maackiain (**5**) [7] by comparison of their spectroscopic data and physicochemical properties with those reported in the literature.

Compounds **1**–**3** belonged to two types of flavonoids, and 2',5'-dioxxygenaurone, 2',5'-dioxxygenchalcone, and  $\beta$ -hydroxydihydrochalcone are all reported here for the first time in the genus *Asragalus*. In fact, they constitute a small subclass of flavonoids found in nature with only a few reports at all. Our findings may be useful for the chemical taxonomy. Since various bioactivities of flavonoids were reported recently, including antioxidant and antiparasitic activities [12–15], the schistosomicidal and antioxidant activities of all separated constituents were tested, and



**Fig. 3** Key  $^1\text{H}$ - $^1\text{H}$  COSY and HMBC correlations of compounds **2** and **3**. (Color figure available online only.)

compounds **2** and **4** showed schistosomicidal activities, while **1** and **2** exhibited antioxidant activities.

All of the isolated compounds (**1**–**5**) were screened for their *in vitro* schistosomicidal and antioxidant activities. Compounds **2** and **4** caused worm mortality rates of 100% within 12 h in a drug-containing (0.70 and 0.77 mM, respectively) RPMI 1640 medium, and the other results are listed in Table 2. Compounds **1** and **2** exhibited stronger antioxidant activities than that of the positive control vitamin C (Vc,  $\text{IC}_{50} = 48.9 \pm 1.1$   $\mu\text{M}$ ), with  $\text{IC}_{50}$  values of  $35.9 \pm 1.1$  and  $12.2 \pm 1.1$   $\mu\text{M}$ , respectively. Compound **3** showed a weak capability ( $\text{IC}_{50} = 871.7 \pm 25.6$   $\mu\text{M}$ ) of DPPH radical scavenging, and compounds **4** and **5** were inactive.

## Materials and Methods

### General

NMR spectra (1D and 2D NMR) were recorded on a Bruker Avance III-400 instrument with TMS as an internal reference. EI-MS and HR-EI-MS data were obtained on a Waters AutoSpec Premier P776 mass spectrometer. IR spectra were recorded by a Bruker Tensor 27 FT-IR spectrophotometer with KBr pellets. Optical rotations were determined on a Jasco P-1020 digital polarimeter, while UV data were obtained on a Shimadzu UV2401PC UV/Vis spectrophotometer. Optical density (OD) values in the antioxidant activity assay were analyzed on a BioTek Synergy HT microplate reader. Silica gel, RP-18, and Sephadex LH-20 were used for open column chromatography. Silica gel GF<sub>254</sub> plates were used for TLC analyses. *Schistosoma japonicum* worms from rabbit hosts were provided by Prof. Yi-Mei Yang's Research Group (School of Basic Medicine, Dali University, China). Gibco RPMI 1640 medium and praziquantel (purity  $\geq 98.0\%$ ) were used on the schistosomicidal activity experiments. 2,2-Diphenyl-1-(2,4,6-trinitrophenyl)hydrazyl (DPPH; purity  $\geq 99.0\%$ ) and Vc (purity  $\geq 98.0\%$ ) were used for the antioxidant activity assay.

### Plant material

Roots of *A. englerianus* were collected in November 2011 from Cangshan Mountain, Dali, Yunnan Province, P.R. China. The plant material was identified by Dr. Chun-Lei Xiang (Kunming Institute of Botany, Chinese Academy of Sciences). A voucher specimen (No. 20100928–1b) has been deposited at the premises of Prof. Bei Jiang's Research Group, Institute of Materia Medica, Dali University.

**Table 2** *In vitro* effects of compounds 1–5 against adult *S. japonicum*.

Group and concentration (mM)		Period of incubation in drug-containing medium								
		12 h		24 h		36 h		48 h		
		MR <sup>a</sup> (%)	VR <sup>b</sup> (%)	MR <sup>a</sup> (%)	VR <sup>b</sup> (%)	MR <sup>a</sup> (%)	VR <sup>b</sup> (%)	MR <sup>a</sup> (%)	VR <sup>b</sup> (%)	
2.5%	DMSO	0	0	0	0	0	0	0	0	
	PZQ	0.12	100	100	66.7	91.7	50.0	87.5	33.3	83.3
	<b>1</b>	0.70	0	50	0	66.7	0	70.8	66.7	87.5
		0.14	0	0	0	0	25	0	25	
	<b>2</b>	0.70	100	100	100	100	100	100	100	
		0.14	0	50	0	50	0	50	50	
	<b>3</b>	0.66	0	25	0	25	66.7	91.7	100	
		0.13	0	0	0	0	0	0	75	
	<b>4</b>	0.77	100	100	100	100	100	100	100	
		0.15	0	0	0	25	0	25	25	
	<b>5</b>	0.70	0	0	0	75	100	100	100	
		0.14	0	0	0	0	25	0	25	

<sup>a</sup> MR: worm mortality rate; <sup>b</sup> VR: worm vigor reduction rate

### Extraction and isolation

The air-dried roots (6.1 kg) were milled and extracted six times with methanol (6 × 25 L) at room temperature. After the extract solutions were combined and concentrated under reduced pressure, the resulting residue (480 g) was suspended in water and partitioned with ethyl acetate and *n*-BuOH successively. The ethyl acetate soluble portion (100 g) was adsorbed onto 100 g of silica gel (80–100 mesh) and chromatographed on a silica gel (1.2 kg, 200–300 mesh) column (diameter × height: 8.5 × 50 cm) eluted with a gradient of CHCl<sub>3</sub>–CH<sub>3</sub>COCH<sub>3</sub> (1:0 to 0:1, 80 L) to yield ten fractions (Frs. A–J). Fr. B (9.4 g) was purified by silica gel column chromatography (CC, 200–300 mesh, 3.2 × 30 cm) with petroleum ether (PE)–acetone (30:1 to 0:1, 10 L) to yield ten fractions (B<sub>1</sub>–B<sub>10</sub>). B<sub>5</sub> (1.6 g) was subjected to repeated CC over silica gel (300–400 mesh, 2.2 × 28 cm) with PE–EtOAc (8:1, 1.5 L) and CHCl<sub>3</sub>–MeOH (240:1, 1 L), then was further purified by Sephadex LH-20 (CHCl<sub>3</sub>–MeOH, 1:1, 0.3 L) and RP-18 CC (1.5 × 12 cm, MeOH–H<sub>2</sub>O, 60% to 100%, 0.5 L) to afford compounds **4** (102.1 mg) and **5** (853.4 mg). Fr. C (5.7 g) was isolated by silica gel CC (200–300 mesh, 3.2 × 25 cm) with PE–acetone (10:1 to 0:1, 7 L) to yield eight fractions (C<sub>1</sub>–C<sub>8</sub>). C<sub>5</sub> (0.8 g) was successively subjected to silica gel CC (300–400 mesh, 1.5 × 28 cm) using CHCl<sub>3</sub>–MeOH (240:1, 0.9 L) as the eluent, Sephadex LH-20 (MeOH, 0.3 L), and repeated silica gel CC (PE–EtOAc, 3:1, 0.5 L and CHCl<sub>3</sub>–MeOH, 30:1, 0.4 L) to yield compounds **2** (98.0 mg) and **3** (67.2 mg). Fr. D (2.4 g) was purified by silica gel CC (300–400 mesh, 2.2 × 30 cm) with PE–EtOAc (3:1, 3 L) to yield six fractions (D<sub>1</sub>–D<sub>6</sub>). D<sub>4</sub> emerged as an orange red sediment after settling for several hours at room temperature, then was repeatedly washed with PE–EtOAc (3:1, 50 mL) to yield compound **1** (37.1 mg). All purified compounds had a degree of purity >96% based on the TLC analyzed (all compounds exhibited one spot both under UV radiation and when sprayed with 8% H<sub>2</sub>SO<sub>4</sub> in methanol) and NMR spectra (the baseline was smooth with trivial impurity peaks).

(*Z*)-2,5'-Dihydroxy-6-methoxyaurone (**1**): orange red needle crystals (pyridine). UV (MeOH) λ<sub>max</sub> (log ε): 254 (4.02), 288 (3.94), 334 (4.36), 415 (3.96) nm; IR (KBr) ν<sub>max</sub>: 3405, 3382, 3074, 3025, 2834, 1666, 1629, 1590, 1573, 1501, 1433, 1300, 1273, 1025, 910, 869, 821, 811 cm<sup>-1</sup>; EI-MS *m/z* (%): 284 [M]<sup>+</sup> (94), 267 (20), 151 (75), 84 (100); HR-EI-MS *m/z*: 284.0674 [M]<sup>+</sup> (calcd. for

C<sub>16</sub>H<sub>12</sub>O<sub>5</sub>, 284.0685); <sup>1</sup>H NMR (400 MHz, C<sub>5</sub>D<sub>5</sub>N) and <sup>13</sup>C NMR (100 MHz, C<sub>5</sub>D<sub>5</sub>N) data in **Table 1**.

2,2',5'-Trihydroxy-4-methoxychalcone (**2**): red amorphous powder. UV (MeOH) λ<sub>max</sub> (log ε): 259 (4.12), 384 (4.35) nm; IR (KBr) ν<sub>max</sub>: 3395, 3387, 3059, 3026, 3020, 2842, 1639, 1611, 1588, 1511, 1486, 1441, 1301, 1235, 1201, 1021, 987, 846, 838, 824 cm<sup>-1</sup>; EI-MS *m/z* (%): 286 [M]<sup>+</sup> (25), 256 (18), 255 (13), 150 (21), 137 (31); HR-EI-MS *m/z*: 286.0849 [M]<sup>+</sup> (calcd. for C<sub>16</sub>H<sub>14</sub>O<sub>5</sub>, 286.0841); <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>COCD<sub>3</sub>) and <sup>13</sup>C NMR (100 MHz, CD<sub>3</sub>COCD<sub>3</sub>) data in **Table 1**.

(β*R*)-2,2',5',β-Tetrahydroxy-4-methoxydihydrochalcone (**3**): yellow white amorphous powder. [α]<sub>D</sub><sup>20</sup>: -18.82 (c 0.13, MeOH); UV (MeOH) λ<sub>max</sub> (log ε): 224 (4.45), 256 (3.89), 282 (3.81), 354 (3.58) nm; IR (KBr) ν<sub>max</sub>: 3405, 3387, 3283, 3080, 3025, 2920, 2874, 2845, 1736, 1672, 1611, 1513, 1463, 1435, 1301, 1261, 1218, 1201, 1182, 1163, 1035, 892, 842, 795 cm<sup>-1</sup>; EI-MS *m/z* (%): 304 [M]<sup>+</sup> (4), 286 (5), 269 (55), 268 (100), 150 (32); HR-EI-MS *m/z*: 304.0995 [M]<sup>+</sup> (calcd. for C<sub>16</sub>H<sub>16</sub>O<sub>6</sub>, 304.0947); <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>OD) and <sup>13</sup>C NMR (100 MHz, CD<sub>3</sub>OD) data in **Table 1**.

### Schistosomicidal assay

A schistosomicidal assay was performed according to a previous method [16,17] (detailed procedure listed in Supporting Information).

### DPPH free radical scavenging capability assay

Antioxidant activity was tested by the DPPH method, which has been previously described [18] (detailed procedure see Supporting Information).

### Supporting information

Detailed protocols for the schistosomicidal assay and DPPH free radical scavenging capability assay, spectroscopic data, and physicochemical properties of **4** and **5**, as well as 1D and 2D NMR, IR, UV, and MS spectra of **1–3** are available as Supporting Information.

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## Conflict of Interest

▼  
The authors declare no conflict of interest.

## References

- 1 Editorial Board of China Herbal, State Administration of Traditional Chinese Medicine. China herbal. Shanghai: Scientific and Technical Publishers; 1999: 341–355
- 2 Chinese Pharmacopoeia Committee. Pharmacopoeia of China. Beijing: China Medical Science Press; 2010: 283–285
- 3 Kunming Institute of Botany, Chinese Academy of Sciences. Flora Yunnanica. Beijing: Science Press; 2006: 701–739
- 4 Delectis Florae Reipublicae Popularis Sinicae Agendae Academiae Sinicae Edita. Flora Reipublicae Popularis Sinicae. Beijing: Science Press; 1997: 78–349
- 5 He ZQ, Findlay JA. Constituents of *Astragalus membranaceus*. J Nat Prod 1991; 54: 810–815
- 6 Hasan N, Osman H, Mohamad S, Chong WK, Awang K, Zahariluddin ASM. The chemical components of *Sesbania grandiflora* root and their anti-tuberculosis activity. Pharmaceuticals 2012; 5: 882–889
- 7 Yoon JS, Sung SH, Park JH, Kim YC. Flavonoids from *Spatholobus suberectus*. Arch Pharm Res 2004; 27: 589–592
- 8 Zhao X, Mei W, Gong M, Zuo W, Bai H, Dai H. Antibacterial activity of the flavonoids from *Dalbergia odorifera* on *Ralstonia solanacearum*. Molecules 2011; 16: 9775–9782
- 9 Huang HQ, Li HL, Tang J, Lv YF, Zhang WD. A new aurone and other phenolic constituents from *Veratrum schindleri* Loes. f. Biochem Syst Ecol 2008; 36: 590–592
- 10 Alaniya MD, Kavtaradze NS, Lavoï S, Pichette A, Mshvildadze VD. Aurone from *Astragalus microcephalus* stems. Chem Nat Compd 2009; 45: 455–456
- 11 Albogami AS, Karama U, Mousa AA, Khan M, Al-Mazroa SA, Alkathlan HZ. Simple and efficient one step synthesis of functionalized flavanones and chalcones. Orient J Chem 2012; 28: 619–626
- 12 Muiva LM, Yenesew A, Derese S, Heydenreich M, Peter MG, Akala HM, Eyase F, Waters NC, Mutai C, Keriko JM. Antiplasmodial  $\beta$ -hydroxydihydrochalcone from seedpods of *Tephrosia elata*. Phytochem Lett 2009; 2: 99–102
- 13 Verma AK, Pratap R. The biological potential of flavones. Nat Prod Rep 2010; 27: 1571–1593
- 14 Harborne JB, Williams CA. Anthocyanins and other flavonoids. Nat Prod Rep 2001; 18: 310–333
- 15 Harborne JB, Williams CA. Anthocyanins and other flavonoids. Nat Prod Rep 1995; 12: 639–657
- 16 Xiao SH, Keiser J, Chollet J, Utzinger J, Dong Y, Endriss Y, Vennerstrom JL, Tanner M. In vitro and in vivo activities of synthetic trioxolanes against major human schistosome species. Antimicrob Agents Chemother 2007; 51: 1440–1445
- 17 Moraes JD, Nascimento C, Lopes POMV, Nakano E, Yamaguchi LF, Kato MJ, Kawano T. *Schistosoma mansoni*: In vitro schistosomicidal activity of pipartine. Exp Parasitol 2011; 127: 357–364
- 18 Zhang LB, Lv JL, Chen HL. Japonicasins A and B, two new isoprenylated flavanones from *Sophora japonica*. Fitoterapia 2013; 87: 89–92